Photoredox-catalyzed α -C-H alkylation of (poly)-hydroxylated substrates mediated by CO_2

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Carbon dioxide (CO₂) is an overproduced, inexpensive and non-toxic gas which can reversibly bind chemical functions with sufficient Lewis basicity. Consequently, it can be used to modify temporarily the reactivity of the corresponding molecules.^[1] For instance, CO₂-promoted photoredox processes have been developed to achieve the α -C-H alkylation of aliphatic amines.^[2] Our investigations in this area focuses on using CO₂ to transiently modulate the reactivity of hydroxylated molecules in order to develop original photoredox-catalyzed transformations. Recently, we have demonstrated that the catalytic and reversible carbonation of aliphatic polyols under CO₂ atmosphere constitutes a unique strategy to perform the α -C-H monoalkylation of these abundant feedstocks by photocatalytic hydrogen atom transfer (HAT).^[3] Experimental and theoretical mechanistic studies have demonstrated an activation resulting from an intramolecular hydrogen bonding between the carbonate and the remaining alcohol.

In the meantime, we have discovered that CO₂ could be used to slow down keto-enol tautomerism in complex radical cascades.^[4] This observation was applied to the design of an unprecedented cascade triggered by photocatalytic HAT and allowing the conversion of cyclopropyl methanol derivatives into elaborate building blocks in one step.



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